

POVOLOTSKAYA, K.L.

~~Obtaining riboflavin preparations from a biological medium. Trudy~~
VNIVI 5:88-95 '54. (MLRA 9:3)

1. Biokhimicheskaya laboratoriya.
(RIBOFLAVIN) (EREMOTHECIUM ASCHBYII)

POVOLOTSKAYA, K.I.; ZAYTSEVA, N.I.

Decomposition of riboflavin by visible light. Trudy VNIIV 5:
145-151 '54. (MLRA 9:3)

1. Biokhimicheskaya laboratoriya.
(RIBOFLAVIN) (PHOTOCHEMISTRY)

BUKIN, V.N.; POVOLOTSKAYA, K.L.; KONDRASHOVA, A.A.; SKOROBOGATOVA, Ye.P.

Fluorometric method for the determination of thiamine. Vit. res. i
ikh isp. no.3:91-99 '55. (MLRA 9:4)

(THIAMINE) (FLUORIMETRY)

POVOLOTSKAYA, K.L.; ZAYTSEVA, N.I.; SKOROBGATOVA, Ye.P.

Fluorometric method for the determination of riboflavin. Vit. res.
1 ikh isp. no.3:108-120 '55. (MIRA 9:4)

(RIBOFLAVIN) (FLUOROMETRY)

POVOLOTSKAYA, K.L.; SKOROBOGATOVA, Ye.P.; ZAYTSEVA, N.I.

Microbiological method for the determination of riboflavin. Vit. rez.
1 ikh isp. no.3:121-128 '55. (MLRA 9:4)

(RIBOFLAVIN) (LACTOBACILLUS CASEI)

POVOLOTSKAYA, K.L.; ZAYTSEVA, N.I.

Chromatographic method for the separation of riboflavin, and its
nucleotides. Vit. res. i ikh isp. no.3:129-132 '55. (MLRA 9:4)

(RIBOFLAVIN) (CHROMATOGRAPHIC ANALYSIS)

POVOLOTSKAYA, K. L.

A method for simultaneous determination of ascorbic acid oxidase, polyphenol oxidase, and peroxidase. K. L. Povolotskaya and D. M. Sedenko (Inst. Plant Physiol., Acad. Sci. U.S.S.R., Moscow). *Biokhimiya* 20, 88-90 (1955).
Two aliquot samples of the tested material are ground in mortars in a pH 5.0 buffer soln. Similarly two other samples are ground in a pH 7.5 buffer soln. The ground material is transferred to 100-ml. volumetric flasks and the mortars are washed with the corresponding buffer solns. The washings are added to the corresponding flasks. The vol. is then made up to 100 ml. The solns. indicated below can be varied depending upon the activity of the enzymes. For the detn. of ascorbic acid oxidase three 100-ml. volumetric flasks were used. Into each of two flasks 4 ml. of the enzyme prepn. of pH 5.0, 2.5 ml. of the buffer soln., and 2.5 ml. of ascorbic acid soln. are added. To the third flask (control) are added 1 ml. CCl_3COOH , 4 ml. of the enzyme prepn., 2.5 ml. of pH 5.0 buffer, and 2.5 ml. of ascorbic acid soln. Flasks are shaken for 30 min. and 1 ml. of 10% CCl_3COOH added. Titration is performed in 50-ml. beakers to which previously were added 5 ml. H_2O and 1 ml. of the material to be tested. The controls are titrated with-out preliminary shaking. For the detn. of polyphenol oxidase six 100-ml. volumetric flasks are used. To 2 are added 2 ml. of the enzyme prepn. of pH 5.0, 2 ml. of corresponding buffer soln., 2.5 ml. of the ascorbic acid soln., and 2.5 ml. of pyrocatechin soln. To 2 other flasks are added 2 ml. of the pH 7.5 enzyme prepn., 2 ml. of corresponding buffer soln., 2 ml. of the ascorbic acid, and 2.5 ml. of the pyrocatechin solns. To each of the two control flasks are added 1 ml. CCl_3COOH , 2 ml. of the enzyme prepn., 2 ml. of the corresponding buffer soln. and 2 ml. of the ascorbic acid and 2.5 ml. of the pyrocatechin solns. Flasks are shaken for 30 min. excepting controls, 1 ml. CCl_3COOH is added to test aliquots, and titrated as above. In this procedure the combined titration value for ascorbic acid oxidase and poly-

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phenol oxidase is detd. From this, the polyphenol oxidase activity value is obtained by subtracting the first titration value from the second. If the titration value for polyphenol oxidase at pH 7.5 is greater than at pH 5.0, the titration value of ascorbic acid oxidase at pH 7.5 must be obtained first and correspondingly subtracted. For the detn. of peroxidase activity three 100-ml. volumetric flasks are used. Into each of 2 are added 1 ml. of the enzyme prepn., 2 ml. of the pH buffer soln., 2.5 ml. of the ascorbic acid soln., 2.5 ml. of pyrocatechin soln., and 1 ml. of H_2O_2 . Into the control flask are added 1 ml. CCl_3COOH , 1 ml. of the enzyme prepn., 1 ml. of the corresponding buffer soln., and 1 ml. each of ascorbic acid, pyrocatechin, and H_2O_2 solns. Shake, acidify, and titrate as above. By this procedure the titration value indicates the combined activity of peroxidase and polyphenol oxidase. By subtracting from this the value obtained above for polyphenol oxidase, the activity of peroxidase is estd. In some instances turbidity may develop in the flasks owing to the complete oxidation of the ascorbic acid. In such cases it is pointless to complete the titration. In all cases the activity of the enzymes is judged by the difference in the no. of ml. of 2,6-dichloroindophenol used up in the control and test titrations. All tests are carried out at room temp. (18-22°).

Temp. fluctuations above and below these limits are to be avoided, and the same end point color intensity should be developed. It was found that in different plant objects the activity of each of the oxidases and their interratios varied depending upon their physiol. state, thereby serving as indexes of direction of the plant tissue energy processes.

B. S. Levine.

H. S. Levine

POVOLOTSKAYA, K. L.

✓ Some changes in metabolism of flowers and plants of tomato after action of 2,4-D and 2,4,5-T. Yu. V. Rakitin, K. L. Povolotskaya, and D. M. Sedenko (E. A. Timiryazev Inst. Plant Physiol., Acad. Sci. U.S.S.R., Moscow). *Fiziol. Rastenii* 3, 297-305(1956).—Application of stimulative doses of 2,4-D and 2,4,5-T to tomato flowers results in increased activity of oxidation-reduction enzymes, increased concn. of P components, particularly protein-lipide P with a higher ratio of org. P to inorg. P. Use of hindering doses of the substances disorganizes the oxidation-reduction systems, usually by enhancement of peroxidase and reduction of polyphenoloxidase systems. Respiration rises mainly in flavo-protein portion. Under these conditions the plant accumulates P, mainly inorg. forms and the ratio of org. to inorg. P declines. G. M. Kozolapov

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Experimental scurvy in relation to the character of the basic diet. M. F. Glazunov and K. L. Pyzdolzhaya. *Russ. Applied Botany, Genetic Plant Breeding* (U. S. S. R.), Suppl. 67, 217-24 (1934). --The Bezsmenov diet, even with the addn. of vitamin C (3 cc. of orange juice), is not satisfactory to prevent scurvy in guinea pigs. Addition of autoclaved hay overcome the deficiencies. Animals kept on the Bezsmenov diet tend to develop fatty degeneration of the liver. This condition does not appear if autoclaved hay is added to the diet.

J. S. Joffe

ASB-55A METALLURGICAL LITERATURE CLASSIFICATION

Sauerkraut as a source of antiscorvy vitamin. K. I.
Pavlova-Skaya, V. N. Bukin, L. A. Gander and M. St.
Makarov. - *Dokl. Akad. Nauk SSSR, Ser. Biol. Sci.*
196 (U. S. S. R.), Suppl. 87, 1031-1032 (1964). J. S. J.

1ST AND 2ND ORDERS																										3RD AND 4TH ORDERS																									
COMMON ELEMENTS																																																			
COMMON VARIANTS																																																			
Testing the anticorruv activity of the preparation of methyl-nor-narcotine. K. L. Ponomareva. <i>Bull. Applied Botany, Genetics Plant Breeding</i> (U. S. S. R.), Suppl. 67, 185-9(1934).—Biol. and chem. tests of 3 preps. of methyl-nor-narcotine have shown that these preps. are devoid of vitamin C. J. S. Joffe																																																			
ASA-SLA METALLURGICAL LITERATURE CLASSIFICATION																																																			

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1ST AND 2ND ORDERS																										3RD AND 4TH ORDERS																									
PROCESSING AND PROPERTY INDEX																																																			
CO ME The antiscorvy vitamin in various fruits, berries and vegetables of the north. V. N. Bukin, K. L. Ponomorenko and N. P. Onokhova. <i>Bull. Appl. Botany, Genetics Plant Breeding</i> (U. S. S. R.) Suppl. 67, 25-31 (1934). --Of the 34 varieties of 8 kinds of vegetables, kohlrabi and red and savoy cabbages proved to have the highest vitamin content. 1 g. acted as a prophylactic for the guinea pig (1 to 4 g. of the ordinary cabbage was required to accomplish similar results). Cultivated strawberries were more antiscorbutic than the wild strawberries. Of the wild fruits the dogrose shows an exceptional potency; 0.1 g. is an effective dosage. The bird cherry, dewberry and Amur wild grape are practically devoid of vitamin C. <i>Prunella pubescens</i>, <i>P. pennsylvanica</i> and <i>P. fuscomaculata</i> are as potent as the orange. J. S. Joffe																																																			
A 10-51.4 METALLURGICAL LITERATURE CLASSIFICATION E 277 72.12.12.1																																																			

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The biological method in determining the anticarcinogenic vitamin. V. N. Bukin, K. L. Povolotskaya and M. F. Glazunov. *Bull. Applied Botany, Genetics Plant Breeding* (U. S. S. R.), Suppl. 67, 195-215(1934).—A number of diets have been investigated with the Bezmonov diet as the standard. The latter was found to be unsatisfactory inasmuch as prophylactic doses of substances containing vitamin C, like orange juice, did not give any response. Addition of irradiated cereals, carotene and autoclaved hay and autoclaved carrots was found to be satisfactory. Carotene in place of carrots was also satisfactory. The latter serves as a source of vitamin D. It also improves the vitamin A content. J. S. Ioffe

ASM-SLA METALLURGICAL LITERATURE CLASSIFICATION

CHEMICAL METHODS IN DETERMINING VITAMIN-C.
V. M. BURNI and K. L. POVOLOKHINA (Bull. Appl.
Sci. U.S.S.R., 1964, Suppl. 67, 179-184).—The
Rumkov reaction and I-thionin do not measure
vitamin-C activity. Tilmann's 2:6-dichlorophenol-
indophenol reaction is not up, but serves for pre-
liminary observations. The reagent is reduced by
sugars at high temp. Ch. Ann. (p)

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Biological method for determining vitamin-D.
V. N. BOMIN, K. L. POKHODNAYA, and M. F.
GRABOVY (Biol. Appl. Bot. USSR, 1964, Suppl.
07, 108-118).—Frequently done of vitamin-D
(e.g., orange juice) do not improve the Bominov
diet. Addition of irradiated carrots, carotene (II),
and beta were also ineffective. A diet of autoclaved
hay and autoclaved carrots (II) was satisfactory.
(I) may replace (II), which serves as a source of -D
and improves the -A content. Ch: Am. (p)

BC

Therapeutic method of determining vitamin-C.
K. L. PYTELOVSKAYA (Bull. Appl. Sci. U.S.S.R.,
1956, Suppl. 5, 225-227). Compared with the
prophylactic, the therapeutic method with guinea-pigs
suffering from scurvy gives sufficiently exact results,
wt. increase and macroscopic pathological changes
being the criteria. The min. healing and preventive
doses are nearly the same. NOTE. Abs. (m)

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100

RAKITIN, Yu.V., otv. red.; ARKHANGEL'SKIY, N.I., red.; KRETOVICH, V.L., red.; METLITSKIY, L.V., red.; SHTEYNBERG, D.M., red. [deceased]; SHCHERBINOVSKIY, N.S., red.; YAKOVLEV, B.V., red.; POVOLOTSKAYA, K.L., red.; SUSHKOVA, L.A., tekhn. red.; VOLKOVA, V.V., tekhn. red.

[Scientific principles in crop protection] Nauchnye osnovy zashchity urozhaya. Moskva, Izd-vo AN SSSR, 1963. 246 p. (MIRA 17:1)

1. Vsesoyuznaya akademiya sel'skokhozyaystvennykh nauk im. V.I.Lenina. 2. Institut fiziologii rasteniy im. K.A. Timiryazeva AN SSSR. Moskva (for Rakitin). 3. Institut evolyutsionnoy fiziologii im. I.M.Sechenova AN SSSR, Leningrad (for Yakovlev). 4. Institut biokhimii im. A.N.Bakha AN SSSR, Moskva (for Metlitskiy).

(Crop yields)

POVOLOTSKAYA, K.L.; RAKITIN, Yu.V.; KHOVANSKAYA, I.V.

Participation of heteroauxins in the translocation of sugars in
plants. Fiziol. rast. 9 no.3:303-308 '62. (MIRA 15:11)

1. K.A.Timiriachev Institute of Plant Physiology, U.S.S.R. Academy
of Sciences, Moscow.

(Indolacetic acid) (Sugars) (Plants--Respiration)

POVOLOTSKAYA, K.L.

Mechanism of the action of maleic hydrazide in plants. Izv. AN
SSSR. Ser. biol. no.2:250-255 Mr-Apr '61. (MIRA 14:3)

1. Timiryazev Institute of Plant Physiology, Academy of Sciences
of the U.S.S.R., Moscow.
(GROWTH INHIBITING SUBSTANCES) (MALEIC ACID)

POVOLOTSKAYA, K.L.

Using gibberellins for stimulating the growth and development of
plants. Itogi nauki: Biol.nauki no.2:383-397 '58. (MIRA 14:4)

(Gibberellin)

POVOLOTSKAYA, K.L.

Thinning flowers and inhibiting the flowering of fruit trees.

Itogi nauki: Biol.nauki no.2:398-415 '58. (MIRA 14:4)

(Fruit culture) (Plants, Effect of chemicals on)

POVOLOTSKAYA, K.L.

Inhibiting the sprouting of potatoes and vegetables in storage.

Itohi nauki: Biol.nauki no.2:416-448 '58.

(MIRA 14:4)

(Potatoes--Storage) (Vegetables--Storage)
(Plants, Effect of chemicals on)

OVCHAROV, K.Ye.; POVOLOTSKAYA, K.L.; ZEMSKAYA, V.A.; SVARINSKAYA, R.A.;
SFDENKO, D.M.

Weed control on fields where broad-leaved plants are grown. Itogi
nauki: Biol.nauki no.2:549-581 '58. (MIRA 14:4)

(Herbicides)

OVCHAROV, K.Ye.; POVOLOTSKAYA, K.L.; ZEMSKAYA, V.A.; SVARINSKAYA, R.A.;
SEDENKO, D.M.

Destroying woody plants and controlling weeds in meadows, pastures,
and nurseries. Itogi nauki: Biol.nauki no.2:582-608 '58.
(MIRA 14:4)

(Herbicides)

POBOLOTSKAYA, K.L.

New form of riboflavin bound with proteins. Biokhimiia 18 no.5:638-643 S-O '53.
(MLRA 6:10)

1. Institut biokhimii im. A.N.Bakha Akademii nauk SSSR, Moscow.
(Riboflavin) (Proteins)

RAKITIN, Yu.V.; POVOLOTSKAYA, K.L.

Microchemical determination of ethylene. *Fiziol.rast.* 7
no.3:366-373 '60. (MIRA 13:6)

1. K.A. Timiryazev Institute of Plant Physiology, U.S.S.R.
Academy of Sciences, Moscow.
(Ethylene) (Fruit—Chemical composition)
(Microchemistry)

MURAV'YEVA, Ye.M., inzh.; POVOLOTSKAYA, M.D., inzh.

Compositions: silver-zinc oxide, and silver-lead oxide obtained
by an oxide annealing method. Elektrotehnika 36 no.3237-39 Mr
'65. (MIRA 1826)

L 51306-65 EPF(c)/ENT(m)/ENP(z)/ENP(i)/ENP(b)/ENA(d)/ENP(t) IJP(c) JD/HB

ACCESSION NR: AP5007535

S/0292/65/000/003/0037/0039

AUTHOR: Murav'yeva, Ye. M. (Engineer); Povolotskaya, M. D. (Engineer) ^{2/3}

TITLE: Silver-zinc oxide and silver-tin oxide compositions obtained by the method of oxidizing annealing ^{21 21 21}

SOURCE: Elektrotehnika, no. 3, 1965, 37-39

TOPIC TAGS: alloy oxidation, silver alloy, zinc containing alloy, tin containing alloy, zinc oxide containing alloy, tin oxide containing alloy, electrical contact alloy.

ABSTRACT: Experiments were performed on samples of silver-zinc and silver-tin alloys in order to obtain compositions of silver with zinc or tin oxide by oxidizing annealing at 800C and to determine the applicability of these compositions to high-power electrical contacts. A composition with oxide in granular form was obtained only from alloys with 3% by weight of either tin or zinc, while on the alloys with 9.7% of tin or zinc a continuous film was formed. Experiments were carried out on samples of cold-rolled alloys, 1.0—2.2 mm thick. Oxidizing annealing was done at 750—800C in oxygen or air. The oxidation of 3% alloys was found to follow the parabolic rate as shown in Fig. 1 of the Enclosure. Alloy density increased very

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L 51306-65

ACCEPTANCE NR: AP5007535

slightly after oxidation but the hardness increased by 100% or more. The variation in resistivity is shown in Fig. 2 of the Enclosure. The contacts were not damaged when subjected to a 2-mm-long arc of about 8 seconds duration. In a corrosion test, samples were exposed to air for 24 hours at room temperature, to distilled water vapor for 20 hours at room temperature and at 100C for 7 hours and then to air at 500C for 5 minutes and at 700C for another 5 minutes. The contact resistance before the test was 0.20 and 0.13 milliohms for silver-zinc and silver-tin alloys, respectively. After the test, the corresponding values were 0.10 and 0.08 milliohms. Contacts made of the alloys were opened and closed 50 times at 60-65 volts and currents of 75, 150, 200, and 300 amperes with no visible damage to the material. The alloys are recommended for further testing in high-power electrical devices. Orig. art. has: 3 figures, 1 formula and 4 tables. [08]

ASSOCIATION: none

SUBMITTED: 00

ENCL: 02

SUB CODE: MM, EE

NO REF SOV: 003

OTHER: 000

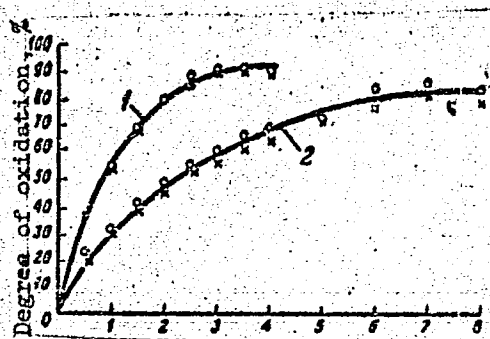
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Card 2/4

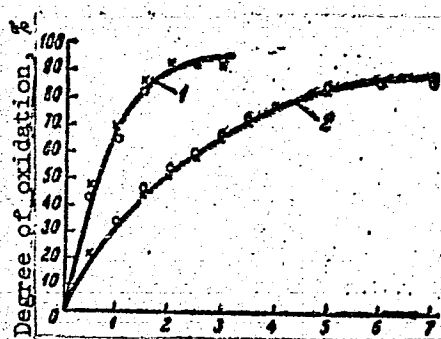
L 51306-65

ACCESSION NR: AP5007535

ENCLOSURE: 01



(a) Oxidation time, hr



(b) Annealing time, hr

Fig. 1. Oxidation curves of the alloys at 800C

a - Silver-zinc (3%); b - silver-tin (3%);
1 - oxygen; 2 - air.

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L 51306-65

ACCESSION NR: AP5007535

ENCLOSURE: 02

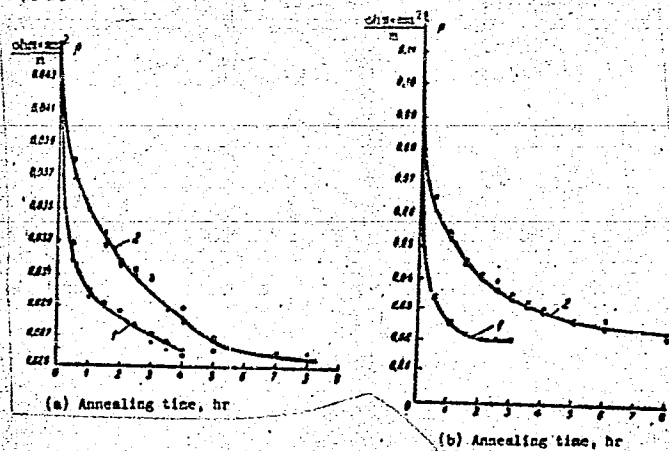


Fig. 2. Variation in alloy resistivity after annealing at 800C

a - Silver-zinc (3%); b - silver-tin (3%);
1 - oxygen; 2 - air.

Card

USOV, Vladimir Vasil'yevich[deceased]; DEKABRUN, I.Ye., red.;
KIRILLOVA, Z.S., red.; POVOLOTSKAYA, M.D., red.; LARIONOV,
G.Ye., tekhn. red.

[Metals for electrical contacts] Metallovedenie elektriche-
skikh kontaktov. Moskva, Gosenergoizdat, 1963. 207 p.
(MIRA 16:6)

(Electric contactors)

PEREDEL'SKIY, M. [Perediel's'kyi, M.]; POVOLOTSKIY, A. [Povolots'kyi, A.];
TELEDIDO, A.; BARANOVSKIY, A. [Baranovs'kyi, A.], glavnyy red.;
DROGICHINSKIY, N. [Drohichyns'kyi, N.], red.; KOCHUBEY, A., red.;
OLEKSYUK, I., red. [deceased]; ZHURBA, S., otv. za vypusk;
LYAMKIN, V., tekhn.red.

[The Soviet Ukraine in the seven-year plan, 1959-1965] Radians'ka
Ukraina v semirichtsi, 1959-1965. Kyiv, Derzhpolitvydav URSR, 1959.
42 leaves. (MIRA 13:5)

(Ukraine--Economic policy)

POVOLOTSKIY, B.

Factory committee permitted. Okhr.truda i sots.strakh. 5 no.11:
29 N '62. (MIRA 15:12)
(Donskoy (Tula Province)---Employees, Dismissal of)

POVOLOTSKIY, D. S., kand. tekhn. nauk

Vibration compaction of fine coking coal transported in gondola cars. Ugol' 38 no.4:43-44 Ap '63. (MIRA 16:4)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut zheleznodorozhnogo transporta Ministerstva putey soobshcheniya.

(Coal handling)

POVOLOTSKIY, D.Ya.

Conditions of flake formations in steel. Izv. vys. ucheb. zav.;
chern. met 6 no.2:120-128 '63. (MIRA 16:3)

1. Chelyabinskiy politekhnicheskii institut.
(Steel—Hydrogen content)

PEREVERZEV, D.A., inzh.; POVOLOTSKIY, L.V., inzh.

Study of the efficiency of a steam turbine rotor cooling system
with supercritical parameters. Ebergomashinost ~~no~~enie 9 no.4:
9-13 Ap '63. (MIRA 16:5)

(Steam turbines—Cooling)

POVOLOTSKIY, Igor' Aleksandrovich; PERLOV, Anatoliy Kuz'mich; CHUGUNIKHIN,
S.I., otv.red.; SHOROKHOVA, A.V., red.izd-va; SABITOV, A., tekhn.
red.

[KN-1 cutter loader] Nareznoi kombain KN-1. Moskva, Gos.nauchno-tekhn.
izd-vo lit-ry po gornomu delu, 1960. 153 p. (MIRA 13:7)
(Coal mining machinery)

POVOLOTSKIY, Ye.G.; DOVGALEVSKIY, Ya.M.; BAYTINA, V.K.

Rate of cooling of "magnico" alloys. Izv.vys.ucheb.zav.; Chern.
met. 6 no.1:120-124 '63. (MIRA 16:2)

1. Saratovskiy politekhnicheskii institut.
(Iron-nickel-cobalt alloys--Magnetic properties)

POVOLOTSKAYA, K. I.

Transformations and functions of a form of riboflavin firmly bound to a protein. Vitaminy no.4:148-153 '59. (MIRA 12:9)

1. Institut fiziologii rasteniy im. K.A.Timiryazeva Akademii nauk SSSR, Moskva.

(RIBOFLAVIN)

(PROTEINS)

ZAYTSEVA, Z.I.; POVOLOTSKAYA, K.L.

Riboflavin released from vegetable proteins isolated by 0,2o/o
NaOh. Dokl. AN SSSR 118 no.2:338-339 Ja '58. (MIRA 11:4)

1. Institut biokhimii im. A.N. Bakha Akademii nauk SSSR i Institut
fiziologii rasteniy im. K.A. Timiryazeva Akademii nauk SSSR.

(RIBOFLAVIN) (PROTEINS)

POVOLOTSKAYA, K. L.

Czechoslovakia/Plant Physiology. Growth and Development I

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58237

Author : Rakitin Yu. V., Povolotskaya K. L.
Inst : Institute of Physiology, Academy of Sciences
USSR

Title : Fluorometric Method of Determination of Heteroauxin in Plants

Orig Pub : Fiziol. rasteniy, 1957, 4, No 3, 285-292

Abstract : On the basis of the Ebert fluorometric method of determination of indolyl acetic acid (IAA) in pure solutions, a method for the determination of IAA in plant tissues has been developed. The fluorescence of the products of the reaction between IAA and sulfuric acid in the presence of copper sulfate was determined on a fluorometer with two light filters. The inten-

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Czechoslovakia/Plant Physiology. Growth and Development I

Abs Jour : Ref Zhur-Biol., No 13, 1958, 58237

Abstract : sity of the fluorescence was proportional to the concentration of IAA within the limits of 0.5 to 2 /ml. Tryptophan, -indolyl butyric acid, -indolyl propionic acid, N-ethyl indolyl-3-acetic acid, -naphthyl acetic acid, DL-catechin, epigallo catechin, chlorophyll, riboflavin, and antocyanins were not fluorescent as a result of the reaction with sulfuric acid and copper sulfate. To determine the concentration of indolyl acetic acid, a standard scale was constructed. The free indolyl acetic acid was extracted with the help of acidified alcohol with ether linked after hydrolysis with an alcohol alkali. The maximum spetrums of absorption in the reaction of indolyl acetic acid and H_2SO_4 with the extract from tissues coincided, a fact which pointed to their identity. The

Card 2/3

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RAKITIN, Yu.V.; POVOLOTSKAYA, K.L.; GHEYDEN, T.M.; GARAYEVA, K.G.

Maleic acid hydrazide as a means of inhibiting the sprouting of
sugar beet roots during prolonged storage.. Fiziol. rast. 5 no.3:
291-295 My-Je '58. (MIRA 11:6)

1. Institut fiziologii rasteniy im. K.A. Timiryazeva Akademii nauk
SSSR, Moskva.

(Sugar beets--Storage)
(Maleic acid)

PUVULLEKAAA, K L

U.S. / 2. Physiology - Developmental - Bone.

2-1

See Also : Ref. 1 - Biol., 11, 4, 1955, 1956

Author : Vekul, K. L., Sovetskaya . . .

Inst : Institute of Human Physiology, Academy of Sciences, U.S.S.R.

Title : Experimental "osteoporosis" in infants, caused by white phosphorus.

Orig. No. : Fiziol. Zhurn. 1957, 4, 11-12, 117-123

Abstract : The treatment of crura and thoracic vertebrae with 0.05%, 0.1, and 0.5% solutions of white phosphorus led to growth inhibition, especially of the costal and side ribs (intensified with an increase in the dose), bending and shortening of the thoracic girdle, and discoloration of the ribs and change of the thoracic cage into yellow-green. Depression of osteoblastic, osteocytic, and osteoclastic activities (especially in the thoracic cage), inhibition

Class 1/2

POVOLOTSKAYA, K.L.

POVOLOTSKAYA, K.L.

Fourth All-Union Conference on Vitamins. Fiziol. rast. 4 no.6:575-576 N-D
'57. (MIRA 10:12)

(Moscow--Vitamins--Congresses)

POVOLOTSKAYA, K.L.

Some coenzymatic functions of riboflavin. Vitaminy no.2:134-143 '56.
(MLBA 10:8)

1. Institut fiziologii rasteniy im. K.A.Timiryazeva Akademii nauk
SSSR, Moskva
(RIBOFLAVIN) (COENZYMES)

X
POVOLOTSKAYA, K.L.

OVCHAROV, K.Ye.; POVOLOTSKAYA, K.L.

Experimental avitaminosis in plants induced by sulfanilamide
[with summary in English]. Fiziol.rast. 4 no.2:117-123 Mr-Apr
'57. (MLRA 10:5)

1. Institut fiziologii rasteniy im. K.A. Timiryazeva Akademii nauk
SSSR, Moskva.

(Sulfanilamide) (Deficiency diseases in plants)
(Vitamins)

POVOLOTSKAYA, K. L.

13

Chemical and spectroscopic methods for determination of vitamin A. I. N. Garkina. *Vitaminnye Resursy i Ikh Ispol'zovanie*, Akad. Nauk S.S.S.R., Inst. Biokhim. i med. A. N. Bakha, *Shornik* 3, 6-21 (1956).—Phys. and chem. properties, structure, and methods of detn. of the A vitamins are reviewed, and details of the chem. and spectroscopic detn. are presented for application to milk, eggs, and blood. Chemical method for determination of a vitamin D. I. N. Garkina and V. N. Bukin. *Ibid.*, 22-52.—Properties of ergosterol, vitamin D₂ and D₃, and 7-dehydrocholesterol, and methods for detn. are reviewed. The chem. detn. methods are presented in detail for analyses of oil and fatty tissue of fish and marine animals, oil solus. of irradiated ergosterol and alc. solns. of irradiated ergosterol. Chromatographic method for separation of provitamin and vitamin D. I. N. Garkina. *Ibid.*, 63-73. Biological method for the determination of the D vitamin activity (line test). N. N. Erofeeva. *Ibid.*, 74-81. Biological method for determining P-vitamin activity. N. N. Erofeeva. *Ibid.*, 82-90. Fluorometric method for determination of vitamin B₁. V. N. Bukin, K. L. Povolotskaya, A. A. Kon-drashova, and E. P. Skorobogatova. *Ibid.*, 91-9. Cat-ionic CDV-3 in the fluorometric method for determination of vitamin B₁. A. A. Dmitrovskii, *Ibid.*, 106-7. Fluoro-metric method for determination of riboflavin. K. L. Povolotskaya, N. I. Zaitseva, and E. P. Skorobogatova. *Ibid.*, 108-20. Microbiological method for determination of riboflavin. K. L. Povolotskaya, E. P. Skorobogatova, and N. I. Zaitseva. *Ibid.*, 121-8.—The method is based on growth of a culture of *Lactobacillus casei*. Chromatographic method for separation of riboflavin and its nucleotides. K. L. Povolotskaya and N. I. Zaitseva. *Ibid.*, 129-32. Microbiological method for determination of nicotinic acid (vitamin PP). O. I. Pushkinskaya and I. S. Kutseva. 1/2

G. N. K. I. N.

Ibid. 123-44. —The method is based on growth of a culture of *Lactobacillus orabiosus*. Microbiological method for determination of pyridoxine (vitamin B₆). N. A. Pomoshchnikova. *Ibid.* 145-51. —The method is based on using a culture of *Saccharomyces hihwigi*. Microbiological method for determination of pantothenic acid. N. A. Pomoshchnikova. *Ibid.* 152-7. Fluorometric method for determination of folic acid. N. A. Andreeva. *Ibid.* 158-65; cf. *C.A.* 48, 5257d. Microbiological method for determination of folic acid. O. I. Pushkinskaya and L. S. Kutseva. *Ibid.* 166-74. Microbiological method for determination of vitamin B₁₂. L. S. Kutseva. *Ibid.* 175-81. Chemical method for determination of vitamin B₁₂. V. N. Buzin, L. Ya. Areshkina, and E. P. Skorobogatova. *Ibid.* 182-7; cf. *C.A.* 49, 6352a. —The method is based on extr. of the vitamin with a dil. (0.25%) soln. of NaNO₂, refining the ext., and spectroscopic or colorimetric detn. of the vitamin in the refined ext. Supplement: Ascorbic acid (vitamin C). Anon. *Ibid.* 183-90. Determination of carotens (provitamin A) according to Murri. Anon. *Ibid.* 191-3; cf. *C.A.* 32, 2554d. *M. M. Piskur*

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POVOLOTSKAYA, K.L.

✓ Destruction of riboflavin under influence of visible light.
K. L. Povolotskaya and N. I. Zaitseva. *Trudy Vsesoyuz.
Nauch. Issledovaniy. Vitamin. Inst.* 5, 145-51 (1954). -- Direct
sunlight rapidly destroys riboflavin at all pH values, with
440 mμ range being most destructive. Nevertheless, low
pH tends to preserve riboflavin better than high pH level.
Riboflavin is moderately stable in diffuse light (natural or
artificial). It is more stable to light in a living cell than in
soln. Ascorbic acid and tea tannin tend to protect ribo-
flavin from destruction to some degree. G. M. K.

2

PHASE I BOOK EMPLOYMENT SOV/4.164

Vasozhiznnyy sversheniye po splavn rezhimov. Izv. Kazan, 1957
Radikal'naya splavnyy tverd... (Sov. Met. i Alloys: Transactions of the
First All-Union Conference on Far-East Alloys) Kazan, Metallurgizdat, 1960
436 p. 3,150 copies printed.

Sponsoring Agencies: Akademiya nauk SSSR, Institut sotsializmi; USSR Embassy in Praha; retailas pri nauchno-tekhnicheskoy knazhote.

Ed.: I.N. Sidorov; Ed. of Publishing House: O.N. Kazayeva; Tech. Ed.: P.G. Isent'yeva.

PURPOSE: This collection of articles is intended for metallurgical engineers, physicists, and workers in the machine-building and radio-engineering industries. It may also be used by students of schools of higher education.

CONCLUSIONS: The collection contains technical papers which were presented and discussed at the First All-Union Conference on Rare-Earth Alloys, held in the Institute of Metallurgy, Academy of Sciences USSR, in November 1957. Results of the work of the USSR, Poland, Czechoslovakia, Hungary, Bulgaria, and other countries on the properties and uses of rare-earth alloys with aluminum, magnesium, iron, titanium, zirconium, niobium and their alloys. The effect of rare-earth metals on properties of magnesium alloys and steels is analyzed. The uses of rare-earth alloys as deoxidizers in steel, aluminum, and titanium alloys are discussed. The effect of the addition of certain elements on the properties of heat-resistant steel is examined and alloys with special physical properties (particularly heat-conducting alloys) are discussed. No personalities are mentioned. Soviet and non-Soviet references are given at the end of the collection.

ALLOTS WITH RAIS-PRAL ADDITIONS

Delella, G.P., G.S. Zilberstein, and M. Nisenzon, L.A. Salasnov, and A.L. Polakova. High-Strength and Heat-Containing Alloys of the Copper-Cobalt-Ni-Fe-Al System	63
Delella, G.P., G.S. Zilberstein, and M. Nisenzon. Investigation of Alloys of the Titanium-Niobium-Aluminum and Titanium-Niobium-Aluminum Systems	34
Mal'nev, M.T., G.P. Ivanov, and Ye. A. Iur'evich. Effect of Rare Metals on the Oxidizability of Titanium and of Some Titanium Alloys	423
Mal'nev, M.T., and Ye. A. Iur'evich. Investigation of Titanium-Aluminum-Titanium Ternary Alloy Systems	545

Rare Metals (Cont.)

50712164

PART III. TITANIUM, VANADIUM, NIOBIUM, ZIRCONIUM AND ALLOYS BASED ON THEM

Belandin, A.A.; Kozlovskaya, A.A. Ethanol as a
Dehydrating Catalyst

Tykhina, M.A., and Ye.M. Savitskiy. Phenolic Alloys

Selwyn, S. L., Z.M. Selwyn, A.O. Widdling, and L.L. Lantry, Electro-
plating With Rhodium

Novoz, T.V., and M.D. Parolitskaya. Electrical Contacts Made of Ethenium

Shumetov, Ye. A. The Possibility of Using Alloys on Tungsten With Rhodium For Making Contacts for Automobile Electrical Equipment

BARON, T. V., and T. M. SAYRE, Property of Vanadium, Niobium, and of
Alloys Based on Them

136

SOV/LIC-58-8-1/26

AUTHORS: Professor Usat, V.V. (Doctor of Technical Science) and
 Povolotskaya, M.D. (Engineer)

TITLE: Rhenium as a Substitute for Tungsten for Contacts (Rendiy
 kak zamenitel' vol'frama dlya kontaktov)

PERIODICAL: Vestnik Elektromyshlennosti, 1958, Nr.8, pp 1-4 (USSR)

ABSTRACT: Previous work, mainly foreign, on the use of rhenium for
 contacts is briefly reviewed. The article then describes
 corrosion and erosion tests made on contacts of rhenium
 and some of its alloys. The specimens of compact and
 porous rhenium and also of alloys of rhenium with tungsten
 were prepared in the rare-alloy laboratory of the
 Institute of Metallurgy of the Academy of Sciences of the
 USSR by Prof. E.M. Savitskiy (Dr. Tech. Sci.) and Gard.
 Tech. Sci. M.A. Tytkina. Contacts of rhenium, tungsten,
 and alloys of rhenium with tungsten, were tested for
 resistance to moist air, high temperatures and contact
 arcs to elucidate the conditions under which non-
 conducting films are formed. The test procedure is
 described and the main test results are tabulated.
 Humidity-oven tests continued for 35 days. Temperature
 tests were run at up to 1,000°C. The appearance of the

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SCV/110-58-8-1/26

Rhenium as a Substrate for Tungsten for Contacts

contacts after exposure to moist air is shown in Fig 1, the appearance of the rhenium contacts being much superior to those of tungsten. Tests on the influence of contact areas were made with the equipment illustrated diagrammatically in Fig 2. Measurements were taken of the arc length when breaking 1.5 A, 110 V d.c. and of the voltage drop across the contacts. It will be seen from the table that the arc was shorter with tungsten than with rhenium but that the rhenium contacts were not damaged. Erosion tests were made on a vibrator with d.c. and 500 c/s a.c.; the current was 1 amp, the voltage 48 V. The tests involved 120,000 operations at a rate of 50 per second. On d.c., rhenium contacts were more than tungsten. Compositions of rhenium and silver were tested for welding under heavy currents; welding did not occur and the contacts continued to operate although they were burned and damaged particularly when the silver content was low. It is concluded that rhenium as a contact material has very similar properties to tungsten but has

Card 2/3

Rhenium as a Substitute for Tungsten for Contacts SOV/110-58-8-1/56

more stable contact-conductivity. Rhenium has good technological properties; it is easily impregnated with silver and copper and can form a basis for contact compositions. Contacts of rhenium and its alloys are easily soldered to copper, brass and steel parts by the procedure used for tungsten contacts. Rhenium contacts can be cut and polished by the methods used for tungsten. The main advantage of rhenium over tungsten is its resistance to atmospheric and tropical corrosion; also that it maintains its contact conductivity after exposure to high temperatures and contact-arcs. It has no other special advantages over tungsten. The properties of tungsten-rhenium alloys as contact materials are intermediate between those of the constituents. There are 3 figures, 1 table, and 5 references, 1 of which is Soviet, 3 English and 1 German.

SUBMITTED: April 18, 1958

Card 3/3 1. Conductors--Materials 2. Rhenium--Applications 3. Rhenium
--Properties

USOV, V.V., doktor tekhn. nauk, prof.: POVOLOTSKAYA, M.D., inzh.

Rhenium as a substitute for tungsten in contacts. Vest. elektroprom.
29 no. 8:1-4 Ag '58. (MIRA 11:3)

(Rhenium)
(Tungsten)
(Electric contactors)

USOV, V.V., doktor tekhn. nauk; BEN'KO, P.A., inzh.; MURAV'YEVA, Ye.M., inzh.;
POVOLOTSKAYA, M.D., inzh.

Silver-nickel contacts for electrical apparatus. Vest. elektroprom,
27 no.8:5-11 Ag '56. (MLRA 10:9)

1. Nauchno-issledovatel'skiy institut Ministerstva elektrotekhnicheskoy promyshlennosti.

(Electric contactors)

32176
S/196/62/000/013/003/018
E194/E155

26-2311

AUTHORS: Usov, V.V., and Povolotskaya, M.D.

TITLE: Rhenium as an electrical contact material

PERIODICAL: Referativnyy zhurnal, Elektrotekhnika i energetika, no.15, 1962, 4, abstract 13 B 25. (In the collection of papers "Reni" (Rhenium). Moscow, AS USSR, 1961, 192-197).

TEXT: A disadvantage of tungsten and molybdenum as contact materials is that they are subject to corrosion and the formation of oxide films. Rhenium approximates to tungsten in its physical and mechanical properties, but has more volatile oxides, which facilitates the self-cleaning of rhenium contacts through heating. Moreover, the electrochemical potential of rhenium is more electro-positive and so it should be more resistant to electrochemical corrosion in contact with other metals. Rhenium has been little studied as a contact material. Contacts made of rhenium, tungsten, rhenium-tungsten and rhenium-molybdenum alloys were studied. Measurements were made of contact resistance after holding at a temperature of 100-110 °C for seven hours and after maintaining

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Rhenium as an electrical contact ... S/196/62/000/013/005/018
E194/E155

for five minutes at 500, 700 and 1000 °C. The contact resistance was measured by the volt-ammeter method with a pressure of 700 g [Abstractor's note: In the original erroneously given as 700 T], a current of 1 A, with a voltage source of 6 V. The contact resistance of rhenium contacts is considerably less than that of tungsten. Because of the volatility of the oxides the contact resistance of rhenium contacts drops sharply after holding at 700 °C. In the case of contacts of tungsten and of alloys of tungsten with 15-20% of rhenium the contact resistance increases at high temperatures and after maintaining at 700 and 1000 °C, conductivity is interrupted. Similar results were obtained on contacts of alloys of molybdenum with 10-40% of rhenium. Tests under tropical conditions (with a relative humidity of 95-98%, maintained for eight hours at a temperature of 45 °C, with the rest of the day at room temperature) demonstrated the good corrosion resistance of rhenium contacts; they remained bright and conductivity did not fail after 35 days' testing. The conductivity of contacts of tungsten, alloys of tungsten-rhenium

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E194/E155

and of molybdenum-rhenium, broke down after 35 days' testing. The length of arc between rhenium contacts (before breaking) is considerably greater than that between tungsten contacts under the same conditions; however, this did not destroy the rhenium contacts. Rhenium contacts were more subject to electrical erosion than tungsten. In many cases it may be justified to replace contacts of pure tungsten by rhenium and its alloys because they resist corrosion and maintain contact conductivity despite high temperatures and contact arcs.
7 references.

[Abstractor's note: Complete translation.]

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POVOLOTSKAYA, M. D.

III. PRODUCTION AND CHARACTERISTICS OF CONTACT MATERIALS

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III. PRODUCTION AND CHARACTERISTICS OF CONTACT MATERIALS

133-58-5-9/31
AUTHORS: Kolosov, M. I., Ayzenshtok, I. Ya., Komissarov, A. I.,
Mysina, G. Ye. and Povolotskaya, M. S.

TITLE: The Influence of the Weight of Ingots on the Quality of
Structural Steels (Vliyaniye vesa slitka na kachestvo
konstruktsionnykh staley)

PERIODICAL: Stal', 1958, Nr 5, pp 411-414 (USSR)

ABSTRACT: An investigation of the possibility of increasing the
weight of ingots of steels 18KhVA, 40KhNMA, 12Kh2N4A
and 30KhGSA from 1.2 and 2.65 t to 4.5 t was carried out.
This increase in weight of ingots was necessary in order
to increase the throughput of the casting pit and blooming
mill as well as to increase the degree of deformation on
rolling profiles of a large cross-section (250 to 300 mm).
The investigation was carried out on eight heats made in
a 30-ton electric furnace. The experimental metal was
teemed into 1.18, 2.65 and 4.5 t ingots. In order to
study the character of crystallisation three ingots of
various sizes from each melt of each of the steels
investigated were selected. After slow cooling and a
softening heat treatment from the twelve selected ingots
axial longitudinal plates were cut. The experimental
ingots of 2.65 and 4.56 (charged hot into soaking pits)

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The Influence of the Weight of Ingots on the Quality of
Structural Steels

were rolled on a blooming mill to a cross-section
250 x 250 mm and then on a mill 800 into semis 140 x
140 mm. Ingots weighing 1.18 t were rolled on a mill
800 into semis 140 x 140 mm. For the studies of the
macrostructure and mechanical properties specimens were
taken from semis 250 x 250 on the following distances
from the top of ingots %:

Ingot 2.65 t 19, 58, 98

Ingot 4.5 t 19, 39, 58, 78, 98

The macrostructure of etched specimens was evaluated
according to MAP-MChM scale. Thermal treatment of
specimens for testing mechanical properties was done
according to MPTU2333-49. The macrostructure of ingots
is shown in Figs. 1-4. The results obtained indicated
that: 1. Macrostructural defects in rolled steels were
caused by defects in the cast structure of ingots.
2. Axial intercrystallite cracks in rolled steel
18KhNVA of a cross-section 250 x 250 from 4.5 t ingots
remain unwelded during rolling in spite of a considerable
degree of reduction (in steel 12Kh2N4A they are welded

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The Influence of the Weight of Ingots on the Quality of
Structural Steels

on both profiles 140 x 140 mm and 250 x 250 mm (from ingots of all weights). 3. The axial porosity and v-shaped cracks in ingots of steels 40KhNMA and 30KhGSA are welded during rolling. 4. The degree of development of segregation outside the central zone of ingots depends on the chemical composition of steel and increases with increasing weight of ingots, but does not exceed the degree permitted by MAP-MChM 1951. From the steels investigated the highest development of the segregation was observed in ingots of steel 30KhGSA. 5. The weight of ingot has no influence on the mechanical properties of steels. 6. The indices of mechanical properties of steels investigated were high with the exception of the top part of 4.5 ton ingot of steel 30KhGSA, where strength and plasticity indices were lower than is required by standards. It is concluded that: 1. Increasing the weight of ingots of 18KhNVA steel from 1.18 to 2.65 ton to 4.5 t is not advantageous, as this deteriorates the macrostructure of metal due to developing axial intercrystallite cracks which are not welded during rolling. 2. Steels 12Kh2N4A

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The Influence of the Weight of Ingots on the Quality of
Structural Steels

and 40KhNMA can be cast into 4.5 ton ingots as their
structure and mechanical properties remain satisfactory.
3. The problem of casting steel 30KhGSA into 4.5 t ingots
requires further investigation.
There are 4 figures.

ASSOCIATION: Chelyabinskiy metallurgicheskiy zavod
(Chelyabinsk Metallurgical Works)

Card 4/4

ACCESSION NR: AP4042188

S/0190/64/006/007/1267/1271

AUTHOR: Terent'yev, A. P.; Mochalina, I. G.; Rukhadze, Ye. G.;
Povolotskaya, Ye. M.

TITLE: Study in the series of polymeric chelates. X. Some
physicochemical studies of polymeric chelates based on thio amide
and poly(thio amide) derivatives of pyridine

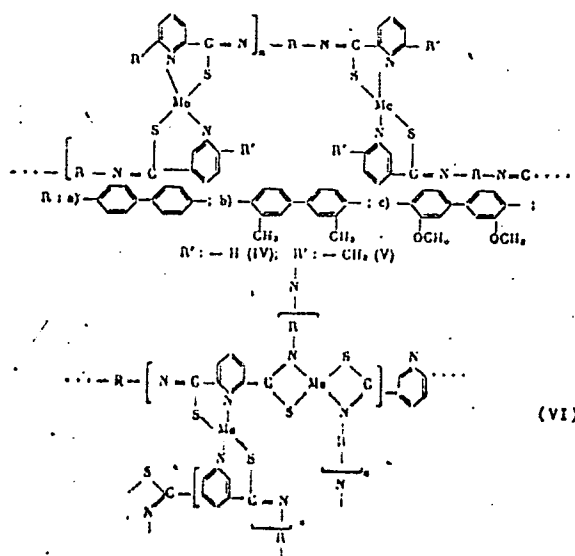
SOURCE: Vy*sokomolekulyarny*ye soyedineniya, v. 6, no. 7, 1964,
1267-1271

TOPIC TAGS: polymeric chelate, thiamide pyridine derivative,
polythiamide pyridine derivative, Cu, Ni, Co, Zn, polymeric chelate
property, chelate group structure

ABSTRACT: The properties of the following Cu-, Ni-, Co-, or Zn-
containing polymeric chelates have been studied.

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ACCESSION NR: AP4042188



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ACCESSION NR: AP4042188

The polymers are fine, colored amorphous powders which are soluble only with difficulty. They dissolve most readily in dimethylformamide, chloroform, and benzene; the best solubility is exhibited by polymers with methoxy groups in the backbone. The polymers are attacked by acids but are resistant to alkalis. They lose 5—15% of their weight on heating to 200C and 20—80% on heating to 400C. Comparative studies showed that polymers with a higher solubility have a lower chemical and thermal stability. The density of polymeric chelates varies from 1.27 to 1.60. They are dielectrics at room temperature; at 383K the highest electrical conductivity (10^{-10} to 10^{-11} ohm $^{-1}$ cm $^{-1}$) is exhibited by Cu-containing polymers. Catalytic properties are inherent only in Cu-, Ni-, and Co-containing polymers. Their catalytic activity surpasses by two orders of magnitude that of inorganic Cu semiconductors. The nature of the band in the chelate group, as determined from x-ray absorption spectra, is -C-S-Me. Orig. art. has: 3 tables.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im M. V. Lomonosova (Moscow State University)

Card 3/4

ACCESSION NR: AP4042188

SUBMITTED: 02Aug63

ATD PRESS: 3073

ENCL: 00

SUB CODE: OC, GC

NO REF SOV: 006

OTHER: 02

Card 4/4

POVOLOTSKAYA, Ye.M.

Case of anaphylactic shock during the use of streptomycin.
Klin. med. 40 no.12:113-114 D '62. (MIRA 17:2)

1. Iz sanatoriya "Pushkino" Moskovskoy oblasti (glavnyy
vrach Z.A. Smirnov).

POVOLOTSKAYA, Yu., vrach (g.Khar'kov)

Long life through rational diet. Obshchestv. pit. no. 3:16-17 Mr '61.
(MIRA 14:4)

(DIET)

ALEKSANDRENKO, Gleb Vasil'yevich [Aleksandrenko, H.V.]; POVOLOTS'KIY,
A., red.; MEYEROVICH, S., tekhn.red.

[Economic councils of the economic administrative regions of
the Ukrainian S.S.R.; legal problems connected with their work
and organization] Radnarhospy ekonomichnykh administratyvnykh
raioniv Ukrain's'koi RSR; pravovi pytannia organizatsii ta
diial'nosti. Kyiv, Derzh.vyd-vo polit.lit-ry URSR, 1959. 158 p.
(MIRA 12:10)

(Ukraine--Economic policy)

POVOLOTSKIY
DARAGAN, M. [Darahan, M.]; SHEVCHENKO, N.; GORIELIK, L. [Horielik, L.],
doktor ekon. nauk, red.; KOROIDA, O., kand. ekon. nauk, red.;
POVOLOTSKIY, A. [Povolots'kiy, A.], red.; KADASHEVICH, O., tekhn.
red.

[Economics of socialist industrial enterprises; an album of
diagrams and plans] Ekonomika sotsialistichnykh promyslovykh pid-
pryemstv; al' bom diagram i skhem. Pid zahal'noiu redaktsiieiu
L. Horielika i O. Koroida. Al'bom uporiadkuvaly M. Darahan i
N. Shevchenko. Kyiv, Derzh. vyd-vo polit. lit-ry URSR, 1958. 46 l.
(in portfolio). (MIRA 11:10)

1. Akademiya nauk URSR, Kiev. Instytut ekonomiky.
(Russia--Economic conditions)

POVOLOTSKIY, A.I. (Leningrad)

Using the variational method for the study of spectra of nonlinear
operators. Mat.sbor. 42 no.3:287-300 J1 '57. (MIRA 10:10)
(Operators (Mathematics)) (Calculus of variations) (Eigenvalues)

~~Povolotskiy, A.I.~~
Povolotskiy, A.I.

USSR .

I - F/W

Povolotskiy, A. I. On the existence of disconnected spectra for nonlinear completely continuous operators. Dokl. Akad. Nauk SSSR (N.S.) 69, 345-348 (1954). (Russian)
Notations: E a Banach space; A a completely continuous operator in E ; Δ - spectrum of

$$A = \{\lambda | Ax = \lambda x, x \neq 0\}; \quad \Pi_{r,B} = \{x | r < |x| < R, x \in E\}, \\ \Delta_{r,B} = \{\lambda | Ax = \lambda x, x \in \Pi_{r,B}\};$$

condition

$$(*) \quad |Ax - Bx| \leq K|x|, \quad x \in \Pi_{r,B},$$

B a linear completely continuous operator;

$$\mathcal{U}_{(\alpha,\beta)} = \{x | x \in \Pi_{r,B}, Ax = \lambda x, \alpha < \lambda < \beta\};$$

$\mathcal{U}_{(\alpha,\beta)}$ is a continuous branch in $\Pi_{r,B}$ if $\mathcal{U}_{(\alpha,\beta)} \cap L \neq \emptyset$ for all

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Toussaint, A. J.

bounded regions L such that $\{x \mid |x| \leq r\} \subset L \subset \{x \mid |x| < R\}$;

$$E_\lambda = \{x \mid Bx = \lambda x\};$$

$$S_{\lambda_m} = \{x \mid \inf \{ \|x - e\| / e \mid e \in E_\lambda \} < m \|x\| \};$$

B_θ = Fréchet derivative of A at θ ; $\lim_{|x| \rightarrow 0} \|Ax - B_\theta x\| / \|x\| = 0$, if B_θ and B_∞ exist. Results (no proofs offered): 1. Let A satisfy condition (*) in Π_{A_m} ; and

$$(**) \quad K < \min \{ \|((\lambda_0 - \epsilon)I - B)^{-1}\|^{-1}; \|((\lambda_0 + \epsilon)I - B)^{-1}\|^{-1} \}.$$

If in $(\lambda_0 - \epsilon, \lambda_0 + \epsilon)$, λ_0 is the only proper value of B and if λ_0 is simple and if B_θ and B_∞ have only one proper value μ , resp. ν , in the same interval, then $\alpha_{\lambda_0 - \epsilon, \lambda_0 + \epsilon}$ is a continuous branch in Π_{A_m} and $\Lambda \cap (\lambda_0 - \epsilon, \lambda_0 + \epsilon) \subset (\mu, \nu)$. 2. When applied to the operator $A\varphi(s) = \int_0^s K(s, t)f(\varphi(t))dt$, where $\int_0^s \int_0^s K^2(s, t) ds dt = 1$, $K(s, t) = K(t, s)$, positive definite, G a

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Provolokii, A. D.

1 - P/W

compact set of measure 1 in n -space, f continuous on $(-\infty, \infty)$, we get: Let $B\varphi(s) = \int_0^s K(s, t)\varphi(t) dt$ have proper values $\lambda_1 > \lambda_2 > \dots > \lambda_n$ which are simple and such that $[\lambda_1, \lambda_n]$ contains no other proper values of B . Furthermore, let (a) $|\mu - f(\alpha)| < 2\alpha$, (b) $f'(0) = 1$, $\lim_{x \rightarrow \infty} f(x)/x = 1 - \alpha$, where $\alpha < \min_{1 \leq j} \frac{1}{2} |\lambda_j - \mu_{j-1}|; 1 - (\mu/\lambda_1)$, where

$$\mu_0 > \lambda_1 > \mu_1 > \lambda_2 > \dots > \lambda_n > \mu_n.$$

Then $(\lambda_0, (1-\alpha)\lambda_1)$ contain different connected components of A . 3. Let A satisfy (2) in $\Pi_{A, B}$, λ_0 the only proper value of B in $(\lambda_0 - \epsilon, \lambda_0 + \epsilon)$, multiplicity α . Let $(**)$ be fulfilled. Then $\lambda_0 \pm \epsilon$ are not in the point spectrum of B , and the sum of the multiplicities of the points in the point spectrum of B , which are in $(\lambda_0 - \epsilon, \lambda_0 + \epsilon)$ is α . Other results in the paper have a similar character.

B. Gelbaum.

$\frac{2}{3}$

POVOLOTSKIY, A.I.

39-3-2/6

AUTHOR:

POVOLOTSKIY A.I. (Leningrad)

TITLE:

Application of a Variation Method for the Investigation of the Spectrum of Nonlinear Operators (Primeneniye variatsionnogo metoda k issledovaniyu spektrov nelineynykh operatorov)

PERIODICAL:

Mat.Sbornik, 1957, Vol.42, Nr.3, pp.287-300 (USSR)

ABSTRACT:

Let Γ be a completely continuous potential operator, $\Gamma = \text{grad } F$, and $F(x)$ a weakly continuous functional. Let $F(\theta) = 0$, $\Gamma \theta = \theta$. Let B be a linear selfadjoint completely continuous operator. If on a set there holds the inequation

$$\|\Gamma x - Bx\| \leq K \|x\|,$$

then the operator Γ is called neighboring to the operator B with the constant K . Let ν be the greatest positive eigenvalue of B , let λ_1 be the second but one eigenvalue. If ν is the only positive eigenvalue, then $\lambda_1 = 0$.

Theorem: Let

$$(1) \quad \|\Gamma x - Bx\| \leq K \|x\| \quad (\|x\| = \xi)$$

and let

$$(2) \quad |F(x) - \frac{1}{2} (Bx, x)| \leq K_1 \|x\|^2 \quad (\|x\| = \xi),$$

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Application of a Variation Method for the Investigation of the Spectrum of Nonlinear Operators 39-3-2/6

where $K_1 < \frac{\nu - \lambda_1}{4}$ and $K < \nu \sqrt{1 - \frac{4K_1}{\nu - \lambda_1}}$. Then on the sphere S ($\|x\| = \rho$), Γ has at least one eigenvector which corresponds to an eigenvalue λ such that

$$|\lambda - \nu| < K + \frac{2(\|B\| + \nu)}{\sqrt{\nu - \lambda_1}} \sqrt{K_1}.$$

Let $\lambda_0 \neq \nu$ be an eigenvalue of B . Let λ_1 be the next smaller eigenvalue (where λ_0 is assumed to be positive).

Theorem: Let (1) and (2) be valid, where

$$K_1 < \frac{\lambda_0 - \lambda_1}{4}$$

and

$$K < \lambda_0 \sqrt{\frac{\lambda_0 - \lambda_1}{\nu - \lambda_1} - \frac{4K_1}{\nu - \lambda_1}}.$$

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Application of a Variation Method for the Investigation of the
Spectrum of Nonlinear Operators

39-3-2/6

Then on S , Γ has at least one eigenvector to which there corresponds an eigenvalue λ such that

$$|\lambda - \lambda_0| < 4K_1 + K.$$

The theorems are proved with the aid of a variation method proposed by Krasnosel'skiy [Ref.2]. In the limit case the author obtains once the theorem of Krasnosel'skiy on the points of bifurcation of a potential operator and another time he obtains the following assertion:

Theorem: Let Γ be the gradient of a weakly continuous functional, which is uniformly differentiable in every sphere of the Hilbert space $\mathcal{H}$. Let the selfadjoint completely continuous operator B_2 be the asymptotic derivative of Γ . Then every eigenvalue of B_2 is an asymptotic eigenvalue of Γ .

Eight Soviet references are quoted.

SUBMITTED: February 1, 1956
AVAILABLE: Library of Congress

Card 3/3

Povolotskiy, A. I.

Call Nr: AF 1108825

Transactions of the Third All-union Mathematical Congress * (Cont.) Moscow,
Jun-Jul '56, Trudy '56, V. 1, Sect. Rpts., Izdatel'stvo AN SSSR, Moscow, 1956, 237 pp.
Povolotskiy, A. I. (Leningrad). On the Structure of Spectrum
of Nonlinear Equation. 118-119

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Operators to Triangular Forms. 119-120

Mention is made of Livshits, M. S.

Sakhnovich, L. A. (Odessa). On the Reduction of Non-self-conjugate
Operators to Diagonal Form. 120-122

Mention is made of Keldysh, M. V., Tamarkin, Ya. D. and
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There are 6 references, 5 of which are USSR, and 1 is English.

Kharazov, D. F. (Tbilisi). Linear Equations With Completely
Continuous Operators, Which are Polynomially Dependent
on Parameters. 122
Card 38/80

*

POVOLOTSKIY, A. I.

"On Spectra of Operators Which Are Nearly Linear." Cand Phys-Math Sci, Chair of Mathematical Analysis, Leningrad State Pedagogical Inst imeni A. I. Gertsen, Leningrad, 1955. (KL, No 18, Apr 55).

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SICH, A.[Sych, A.]; TELEDIDO, A.; TESLYA, P.; CHEGORYAN, O.[Chehorian, O.];
POVOLOTSKIY, A.I.[Povclots'kyi, A.I.], red.; LYAMKIN, V., tekhn.
~~red.~~

[New developments on the map of the Ukraine; album of diagram-
matic maps of economic administrative regions] Nove na karti
Ukrainy; al'bom kartoskhem ekonomichnykh administratyvnykh raioniv.
n.p. Derzhpolitvydav URSR, 1961. 14 maps. (MIRA 15:7)
(Ukraine--Maps)

KRASNOSEL'SKIY, Mark Aleksandrovich; PEROV, Anatoliy Ivanovich;
POVOLOTSKIY, Abram Isaakovich; ZABREYKO, Petr
Petrovich; GORYACHEV, M.M., red.; AKSEL'ROD, I.Sh.,
tekhn. red.

[Vector fields on a plane] Vektornye polia na ploskosti.
Moskva, Fizmatgiz, 1963. 245 p. (MIRA 16:11)
(Vector analysis)

POVOLOTSKIY, A.I.[Povolots'kyi, A.I.], red.; LEVCHENKO, O.K., tekhn. red.

[Soviet Ukraine; figures and diagrams]Radians'ka Ukraina; tsyfry
i diagramy. Kyiv, Derzh. vyd-vo polit. lit-ry, USSR, 1962. 125 p.
(MIRA 16:1)

(Ukraine--Statistics--Charts, tables, etc.)

DAYEN, Leonid Abramovich; FOZNYAK, Pavel Ivanovich; CHERP, Mark
Makovich; POVOLOTSKIY, A.I. [Povolots'kyi, A.I.], red.;
LEVCHENKO, O.K., tekhn. red.

[Kiev; a guidebook and manual] Kyiv; putivnyk-dovidnyk.
Kyiv, Derzh.vyd-vo polit.lit-ry URSS, 1963. 187 p.
(MIRA 17:3)

DAYEN, Leonid Abramovich; POZNYAK, Pavel Ivanovich; CHERP, Mark
Maksovich; POVOLOTSKIY, A.I.[Povolots'kyi, A.I.], red.

[Kiev; reference guidebook] Kyiv; putivnyk-dovidnyk.
Kyiv, Politvydav Ukrainy, 1965. 187 p.
(MIRA 19:1)

MAR'YANOV, B.M.; SICH, A.S.[Sych, A.S.]; YAMPOL'SKIY, B.B.[Iampol's'kyi, B.B.]; VELICHKA, I.O.[Velychka, I.O.], red.; POVOLOTSKIY, A.I. [Povolots'kyi, A.I.], red.; GAVRILETS', D.V.[Havrylets', D.V.], tekhn. red.

[Great 20 years; visual aid]Pro velyke dva~~ts~~tsiatyrichchia; na-
ochnyi posibnyk. Kyiv, Derzhpolitvydav URSR, 1962. 62 p.
(MIRA 16:2)

(Russia—Economic policy)

POVOLOTSKIY, A.I.

Determination of the angular order of a locally simple curve. Dokl.
AN SSSR 124 no.3:524-526 Ja '59. (MIRA 12:3)

1. Predstavleno akademikom F.S. Aleksandrovym.
(Curves)

16(1)

AUTHOR: Povolotskiy, A.I.

SOV/140-59-4-17/26

TITLE: On Eigenvectors on the Hyperboloids in the Hilbert Space

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Matematika, 1959, Nr 4, pp 126 - 140 (USSR)

ABSTRACT: Let B be a linear selfadjoint positive definite completely continuous operator in the real Hilbert space $\mathcal{H}$. Let $\mathcal{H}_1$ be a finite-dimensional invariant subspace of the operator B , $\mathcal{H}_2$ the orthogonal complement of $\mathcal{H}_1$. Let a unitary operator J be introduced which is identical with the unit operator I on $\mathcal{H}_1$ and equal to $-I$ on $\mathcal{H}_2$. The operator J is the gradient of $\frac{1}{2}(Jx, x)$. The equipotential surfaces of this functional are denoted as hyperboloids in $\mathcal{H}$. Let $\bar{\Gamma}$ be a completely continuous potential operator, i.e. $\bar{\Gamma} = \text{grad } F(x)$, where $F(x)$ is a weakly continuous functional uniformly differentiable in $T(\|x\| < R)$. Let $F(\theta) = 0$, $\bar{\Gamma}\theta = \theta$. Theorem: Let in $\bar{\Gamma}_{r,R}$ ($r \leq \|x\| \leq R$) be

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On Eigenvectors on the Hyperboloids in the Hilbert Space

SOV/140-59-4-17/26

$$(2) \quad \|Jx - Bx\| \leq K \|x\| \quad \text{and}$$

$$(3) \quad |F(x) - \frac{1}{2} (Bx, x)| \leq K_1 \|x\|^2$$

where it is $K < \nu m$, $K_1 < \frac{\nu}{4} m^2$, where ν is the smallest positive eigenvalue of JB and $m = \frac{r}{R}$ ($m < 1$). Then on every

hyperboloid $\gamma \left[(Jx, x) = \eta^2 \right]$, where $r^2 \leq \eta^2 \leq \frac{\nu - 4K_1}{\nu + 4K_1} R^2$,

$J\Gamma$ possesses an eigenvector to which there corresponds an eigenvalue λ such that

$$|\lambda - \nu| < 2K_1 + \frac{1}{m^2} (K + 6K_1).$$

Let λ_1 be the smallest of the positive eigenvalues of JB, which are larger than λ_0 ($\lambda_1 > \lambda_0 > \nu$).

Card 2/4

On Eigenvectors on the Hyperboloids in the
Hilbert Space

SOV/140-59-4-17/26

Theorem : Let (2) and (3) be satisfied in $\Gamma_{r,R}$, where

$K < \nu m$ and $K_1 < \min \left\{ \frac{\nu}{4} m^2 ; \frac{\lambda_1 - \lambda_0}{4} m^2 \right\}$. Then on every hyper-

boloid $\gamma \left[(Jx, x) = \eta^2 \right]$, where it is $r^2 \leq \eta^2 \leq \frac{\nu - 8K_1}{2\lambda_0 - \nu} R^2$, $J\Gamma$

possesses an eigenvector to which there exists an eigenvalue λ

such that $|\lambda - \lambda_0| < \frac{1}{2} (K + 4K_1)$.

Theorem : Let $\Gamma(\Gamma \theta = \theta)$ be a non-linear completely continuous operator which is the gradient of a weakly continuous functional being uniformly differentiable in a neighborhood of θ . In the zero point θ let Γ possess a Frechet derivative B_1 which is a self-adjoint positive definite completely continuous operator. Let J_1 be a unitary operator commuting with B_1 ,

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where JB_1 possesses finitely many positive eigenvalues. Then every positive eigenvalue of JB_1 is a bifurcation point of $J\Gamma$ (see Krasnosel'skiy [Ref 3,9_7]).

Theorem : Let B_2 be self-adjoint, completely continuous, positive definite and be asymptotic derivative of Γ , where Γ is the gradient of a weakly continuous functional being uniformly differentiable in every sphere of $\mathcal{H}$. Let the unitary operator J commute with B_2 , JB_2 is assumed to have finitely many positive eigenvalues. Then every positive eigenvalue of JB_2 is an asymptotic eigenvalue of $J\Gamma$ (see Krasnosel'skiy [Ref 3,9_7]).

The author mentions Yu.G.Borisovich, E.S. Tsitlanadze and L.A. Lyusternik. - There are 9 Soviet references.

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Determination of the Angular Order of a Locally Simple Curve
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At the suggestion of M.A. Krasnosel'skiy the author investigates the connection between the order $q(a)$ of a curve with respect to the point a which does not lie on the curve and the angular order p of the curve. Since $q(a)$ does not change under variation of a , if a does not intersect the curve itself, the author speaks of the order of the curve with respect to a domain. The curve Γ is assumed to divide the z -plane into the m simply connected domains D_j ($j = 1, 2, \dots, m$). The corresponding orders then are $q(D_j)$. Let the angular order of Γ be p . Let several branches of Γ pass through the point A ; the branches with the same tangent are denoted as a bundle; let the number of the bundles be k . Two branches each determine a sector of a sufficiently small neighborhood of A . Every sector S_r belongs

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